Cyclopentadienyl Pyridazines and Oxazines and Their Applications in Energy and Advanced Electronics

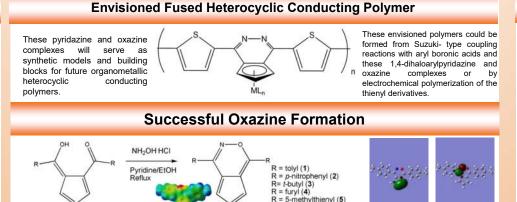


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Introduction/Background

Chemistry Program

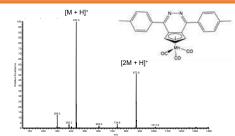
Heterocycles and their derivatives have long been of interest for use in advanced electronic applications. Due to their unique properties including environmental stability, high processibility, and low production cost, organic-based electronic materials are an attractive alternative to conventional inorganic semiconductors. The growing interest and demand for alternative energy and next generation electronic devices has led to the incorporation of aromatic heterocycles into field-effect transistors (FETs), organic lightemitting diodes (OLEDs), and organic photovoltaic (OPV) cells. Two such classes of heterocycles, pyridazines and oxazines (6 membered aromatic rings containing two adjacent nitrogen atoms or an adjacent nitrogen and oxygen), represent alluring building blocks for electronic materials due to their ease of synthesis and stability. However, very little is known about the materials properties of pyridazines and oxazines or the feasibility of incorporating them into electronic devices. Our current investigation focuses on the off-metal and transition metal chemistry of Cp pyridazines and oxazines, namely, the formation of a fused-ring complexes bound n⁵ (eta-5) to a metal through the Cp moiety. By including a transition metal into the pyridazines and oxazines, we hope to create hybrid materials which blend the synthetic versatility of organics with the novel structural and electronic properties that inorganic moieties possess.



Formation of the 1,2-disubstitutedcyclopentadienes (fulvenes) was accomplished by reaction of 3 equivalents of lithium cyclopentadienide and 2 equivalents of the corresponding acid chloride. Ring closure to the oxazine was accomplished by refluxing with hydroxylamine hydrochloride in a pyridine/EtOH mixture, in modest yields (29-46%). X-Ray crystallographic analysis of the tolyl case confirms the desired 5,6-fused oxazine, displaying π -stacking between adjacent layers related by an inversion symmetry element. Natural Bonding Orbitals (NBOs) were calculated for the tolyl case, with the HOMO primarily residing on the Cp portion of the target molecule and the LUMO primarily on the 6-membered oxazine moiety.

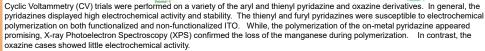
Tice, N. C.; Collins, E. M.; Smith, D. L.; Snyder, C. A.; Yan, B.; Stevens, E.; J. of Heterocyclic Chem., 2017, 54, 3235-3240. Tice, N. C.; Wild, S.; Olmstead, C.; Stevens, E. D.; Yan, B.; Brooks, H.; Jenkins, J. L. Heterocycles 2019, 98, 1707-1724.

DART Analysis of Pyridazine Complexes



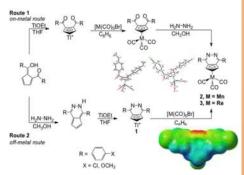
Direct Analysis in Real Time (DART) Mass Spectrometry analysis was shown to be effective for the thallium, manganese, and rhenium pyridazine complexes. This mild and convenient form of MS is better able to confirm structure and molecular weight versus more traditional methods. High relative abundance of the protonated molecular ion peak and dimer was observed under positive mode.

Smith D. L., et. al. Analytical Methods. 2015, 7, 4003-4007



Cyclic Voltammetry Polymerization Trials

Comparison of "On-" vs. "Off-Metal" Routes



Formation of these 5,6-fused ring pyridazines could be accomplished through either a ring-closure reaction on-metal with the corresponding diacyl metallocene (Route 1) or by closure before metallation, followed by complexation with a thallium salt (Route 2). The off-metal synthetic pathway employed (Route 2) did improve upon the isolation of these complexes as compared to previously reported route, with slightly higher yields and better analytical purity.

Tice, N. C.; Snyder, C. A.; et. al. Transition Met. Chem. 2012, 37, 141-147. Snyder, C. A.; Maddox J. B.; Tice, N. C. et. al. Transition Met. Chem. 2013, 38, 801-809.

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